## **COURSE INFORMATION SHEET**

**INSTITUTION:** South Plains College

**DIVISION:** Arts and Sciences

**DEPARTMENT: Science** 

**DISCIPLINE**: Chemistry

**COURSE NUMBER: CHEM 1406** 

**COURSE SECTION: 004** 

**COURSE TITLE**: Introductory Chemistry I

**SEMESTER:** Fall 2017

CREDIT: 4 LECTURE: 3 LAB: 3

## **CORE OBJECTIVES TO BE ADDRESSED:**

**Communication** – to include effective written, oral and visual communication

**Critical Thinking Skills** – to include creative thinking, innovation, inquiry, and analysis, evaluation and synthesis of information

**Empirical and Quantitative Skills** – to include the manipulation and analysis of numerical data or observable facts resulting in informed conclusions

**Teamwork Skills** – to include the ability to consider different points of view and to work effectively with others to support a shared purpose or goal

**PREREQUISITES:** None

**INSTRUCTOR:** John Heh

Office: S-117A

Telephone: 806.716.2323

E-mail: jheh@southplainscollege.edu

**OFFICE HOURS:** Tuesdays and Thursdays 9 AM to 11 AM and 1 PM to 2 PM

Fridays 9 AM – 12 PM

Other times by appointment.

**TUTORS:** More information will be given as it becomes available.

**COURSE MATERIALS:** All are available at the SPC bookstore.

- 1. Karen C. Timberlake, Chemistry: An Introduction to General, Organic, and Biological Chemistry. Thirteenth Edition.
- 2. CHEM 1406 Lab Manual
- 3. Safety glasses or safety goggles (required)
- Scientific calculator [graphing calculators are allowed] (required)
  Cell phones may not be used as a calculator
- 5. 4 Scantron Sheets: Apperson Form 29240
- 6. Mastering Chemistry Code (Instructor will provide that)

**COURSE DESCRIPTION**: Survey course introducing chemistry. Topics may include inorganic, organic, biochemistry, food/physiological chemistry, and environmental/consumer chemistry. Designed for allied health students and for students who are not science majors. Basic laboratory experiments supporting theoretical principles presented in lecture; introduction of the scientific method, experimental design, data collection and analysis, and preparation of laboratory reports. Semester Hours: 4 Lecture Hours: 3 Lab Hours: 3 Note: This course may not be substituted for CHEM 1411.

**COURSE PURPOSE**: To provide basic chemical knowledge for persons living in a world of technology that is always changing. To provide understanding of the basic chemical functioning of the human body. To provide the student with a laboratory experience which will enhance their appreciation of the advances of Science and of the role of the clinical laboratory in the hospital.

**LECTURE EXAMS:** There will be three lecture exams covering the material that is discussed during the lecture portion of this course. The schedule for the lecture exams is given in this course information sheet. Each lecture exam will count 100 points. The lecture exams will be approximately 20 to 40 questions. The format will be multiple choice. A scantron is required for the lecture exams. No outside material may be used on the lecture exams. You may only reference what is provided to you on the lecture exams. You will have the 75 minute designated class time to finish the exam. There will be no make-ups for lecture exams. A missed lecture exam will receive a grade of zero.

Exam 1 (Chapters 2, 4, 6): 100 points Exam 2 (Chapters 3, 7, 8): 100 points Exam 3 (Chapters 9, 10, 11): 100 points

The material scheduled for each lecture exam is subject to change. Changes will be announced if necessary.

**HOMEWORK:** Homework will be assigned often. The homework will not be collected and graded. It is essential that the homework be completed, as the homework will be very similar to the types of problems encountered on the exams.

**LAB EXPERIMENTS:** The laboratory portion of this class will be comprised of topic discussion, homework problems practice; and most commonly, lab experiments. The lab portion of this course will consist of group work. If a student is absent for a lab, the lab cannot be made up.

**LAB SAFETY:** The chemistry laboratory is a potentially hazardous environment. Therefore, all students must follow all of the safety rules passed out to you during the safety presentation. The students must also follow any specific safety rules listed in the lab manual and any ones that the instructor may announce during a lab period. A student not following the safety rules may be asked to leave the laboratory.

LAB GRADE: The lab grade will come from lab reports. For most lab days, we will have a lab report that covers the material accomplished that day in lab. The lab reports will be completed in groups and the due date for the lab reports will be announced during lab. Most lab reports will be due the same day the lab is performed. The lab groups will consist of 2 to 4 students. The lab experiments must be completed on the day that they are scheduled. There will be no make-ups for the lab experiments or the lab reports. If a student misses a lab report, a grade of zero will be assigned for that lab report. The lab reports count 10 points each. There will be ONE lab report turned in per group. Make sure your name is placed on the lab report when it is turned in. If your name is not on the lab report, you will receive a grade of zero. There will be fourteen graded lab reports. The format will mostly be multiple choice. The lowest four lab report grades will be dropped. Therefore, 10 lab reports will count for a total of 100 points.

Lab Reports (10 points each)

100 points total

The material scheduled for each lab is subject to change. Changes will be announced if necessary.

**EXTRA CREDIT:** Extra credit will be available for each exam. The extra credit questions will be on Mastering Chemistry and will require use of the code I will provide you. More information about each assignment will be provided later in the semester.

**FINAL EXAM:** The final exam will not be comprehensive. There will be no make-up exam for the final exam. A missed final exam will receive a grade of zero. The time of the final exam is

given in this course information sheet. The final exam will count 100 points. The final exam will be approximately 25 to 100 questions. The format will be multiple choice. A scantron is required for the final exam. No outside material may be used on the final exam. You may only reference what is provided to you on the final exam. You will have the 120 minute designated final exam time to finish the final exam.

## Final Exam (CARBOHYDRATES, LIPIDS, PROTEINS): 100 points

The material scheduled for the final exam is subject to change. Changes will be announced if necessary.

**FINAL EXAM GRADE:** If the final exam grade is higher than your lowest lecture exam grade, the final exam grade will replace that one lecture exam grade. In that case, the final exam will count twice: once for the final exam grade and once to replace that one lecture exam grade. (e.g. You miss a lecture exam and receive a zero. You later on make an 80 on the final exam. Then the 80 will replace the zero and the 80 will be the final exam grade as well.) This only applies to one lecture exam grade even if multiple lecture exam grades are tied for the lowest lecture exam grade. In all other cases, the final exam counts once and no lecture exam grades are dropped.

ATTENDANCE: Students are expected to attend all classes in order to be successful in this course. A student will be administratively withdrawn from this course when absences become excessive. If you reach a total of 7 absences, you will be dropped from the course with a grade of X. It is the student's responsibility to verify administrative drops for excessive absences through MySPC using his or her student online account. This is in accordance with the policies set forth in the SPC General Catalog. Attendance will usually be taken at the end of the lab period. Absences will be attributed to the student from the first class meeting. This course information sheet contains the schedule of lectures and labs. If you are unable to finish this course, complete a withdrawal slip at the registrar's office. Absences caused by official South Plains College activities will be excused. No other absences will be excused.

**CLASSROOM CONDUCT:** Students are expected to maintain a pleasant learning environment for themselves as well as for their classmates. Therefore, if, in the view of the instructor, a student is disrupting the class, the appropriate disciplinary action may be taken.

**ACADEMIC INTEGRITY:** Cheating (as defined in the SPC General Catalog) will not be tolerated. If a student is caught cheating on an exam, a grade of zero will be given for that exam.

**FINAL COURSE GRADE:** At the end of the semester, all of your points earned will be added together. Your final course grade will come from your point total. The point totals and their corresponding final course grades are listed below:

Point total:	Final Course Grade:		
445 and above	Α		
395 – 444	В		
345 – 394	С		
295 – 344	D		
0 – 294	F		

**DIVERSITY STATEMENT:** In this class, the teacher will establish and support an environment that values and nurtures individual and group differences and encourages engagement and interaction. Understanding and respecting multiple experiences and perspectives will serve to challenge and stimulate all of us to learn about others, about the larger world and about ourselves. By promoting diversity and intellectual exchange, we will not only mirror society as it is, but also model society as it should and can be.

**SPC STANDARD DISABILITY STATEMENT:** Students with disabilities, including but not limited to physical, psychiatric, or learning disabilities, who wish to request accommodations in this class should notify the Disability Services Office early in the semester so that the appropriate arrangements may be made. In accordance with federal law, a student requesting accommodations must provide acceptable documentation of his/her disability to the Disability Services Office. For more information, call or visit the Disability Services Office at Levelland (Student Health & Wellness Office) 806-716-2577, Reese Center (Building 8) & Lubbock Center 806-716-4675, or Plainview Center (Main Office) 806-716-4302 or 806-296-9611.

**NON-DISCRIMINATION STATEMENT:** South Plains College does not discriminate on the basis of race, color, national origin, sex, disability or age in its programs and activities. The following person has been designated to handle inquiries regarding the non-discrimination policies: Vice President for Student Affairs, South Plains College -1401 College Avenue, Box 5, Levelland, TX 79336, 806-894-9611

**Note to students with disabilities:** If you have a disability-related need for reasonable academic adjustments in this course, provide the instructor with a letter of accommodation from the Disability Services Office. If you need immediate accommodations or physical access, please arrange to meet with the Disability Services Office before the next class meeting.

**SAFETY RULES:** These safety rules will be passed out in lab. The safety rules must be followed. Failure to do so can result in you being asked to leave the laboratory. You will be required to sign a sheet indicating you have read and agreed to follow the safety rules before being allowed to perform an experiment.

**COURSE SCHEDULE:** The following table contains the tentative course schedule. *All material* (including lecture material, experiment material, and material scheduled for the lecture exams) is subject to change. Also, all dates are subject to change. Changes will be announced if necessary.

COURSE SCHEDULE						
WEEK	DAY 1 - LECTURE	DAY 1 - LAB	DAY 2 - LECTURE	DAY 2 - LAB		
1	CHAP 2: UNITS, CONVERSIONS, DENSITY	NO LAB	CHAP 2: UNITS, CONVERSIONS, DENSITY	NO LAB		
2	NO CLASS	NO LAB	CHAP 2: UNITS, CONVERSIONS, DENSITY	SAFETY RULES, EXP 1: LAB EQUIPMENT		
3	CHAP 4: ATOMS AND ELEMENTS	EXP 2: MEASUREMENTS	CHAP 4: ATOMS AND ELEMENTS	EXP 3: DENSITY		
4	CHAP 6: BONDING AND NOMENCLATURE	EXP 5: ATOMS AND MOLECULES	CHAP 6: BONDING AND NOMENCLATURE	PRACTICE		
5	CHAP 6: BONDING AND NOMENCLATURE; CHAP 3, 8: ENERGY AND GASES	EXP 6: MODELS	EXAM 1 (CHAP 2, 4, 6)	NO LAB		
6	CHAP 3, 8: ENERGY AND GASES	EXP 9: BOYLE'S LAW	CHAP 3, 8: ENERGY AND GASES; CHAP 7: CHEMICAL REACTIONS	EXP 4: HEAT OF FUSION		
7	CHAP 7: CHEMICAL REACTIONS	PRACTICE	CHAP 7: CHEMICAL REACTIONS	EXP 7: MOLE RATIOS		
8	CHAP 9: SOLUTIONS AND CONCENTRATION	PRACTICE	EXAM 2 (CHAP 3, 7, 8)	NO LAB		
9	CHAP 9: SOLUTIONS AND CONCENTRATION	EXP 8: HYDRATES	CHAP 10: ACIDS, BASES, pH	PRACTICE		
10	CHAP 10: ACIDS, BASES, pH	EXP 10: SOLUTIONS	CHAP 10: ACIDS, BASES, pH; CHAP 11: ORGANIC COMPOUNDS	EXP 11: HOUSEHOLD ACIDS AND BASES		
11	CHAP 11: ORGANIC COMPOUNDS	EXP 12: TITRATION	CHAP 11: ORGANIC COMPOUNDS	EXP 13: ORGANIC MODELS		
12	CARBOHYDRATES	PRACTICE	EXAM 3 (CHAP 9, 10, 11)	NO LAB		
13	CARBOHYDRATES	NO LAB	NO CLASS	NO LAB		
14	LIPIDS	NO LAB	PROPERTIES OF PROTEINS	NO LAB		
15	PROPERTIES OF PROTEINS	NO LAB	REVIEW	NO LAB		
16	CHEM 1406.004: FINAL	EXAM (CARBOHYDRATE	S, LIPIDS, PROTEINS): DEC 11 - 1:00	0 PM - 3:00 PM		